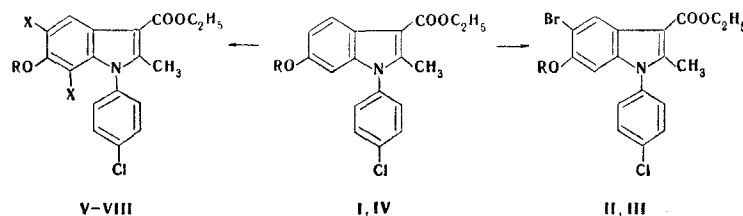


ELECTROPHILIC SUBSTITUTION IN 6-HYDROXYINDOLES

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We were able to demonstrate that 5-substituted (during bromination) or 5,7-disubstituted (during nitration) indoles are formed during electrophilic substitution (nitration and bromination) reactions of both 6-methoxy- [1] and 6-acetoxyindoles in which the pyrrole ring is substituted. Thus 1-(p-chlorophenyl)-2-methyl-3-carbethoxy-5-bromo-6-methoxyindole (II) with mp 179-181° was obtained in 71% yield in the bromination of 1-(p-chlorophenyl)-2-methyl-3-carbethoxy-6-methoxyindole (I).



I, II R=CH₃; IV, V, VII R=H; III, VI, VIII R=CH₃CO; V, VI X=Br; VII, VIII X=NO₂

The PMR spectrum of the product in trifluoroacetic acid contained signals of the 4-H and 7-H protons (8.07 and 6.50 ppm) and a quartet of protons of the p-chlorophenyl ring (7.47, 7.32, 7.14, and 6.99 ppm; J=9 Hz). 1-(p-Chlorophenyl)-2-methyl-3-carbethoxy-5-bromo-6-acetoxyindole (III) with mp 140-142° was similarly obtained in 86% yield. The signals of the 4-H and 7-H protons and the quartet of protons of the p-chlorophenyl ring are observed in the PMR spectra. The bromination of a compound with an unsubstituted OH group (IV) gave 5,7-dibromoindole (V) with mp 187-189° in 66% yield. Its acetate (VI) (mp 167-169°) has the singlet of a 4-H proton (8.18 ppm) and the quartet of protons of the p-chlorophenyl ring in its PMR spectrum. 5,7-Dinitroindole (VII) with mp 208-210° is obtained in yields up to 38% under various conditions in the nitration of this same compound (IV). Its acetyl derivative (VIII) (mp 165-166°) has the signal of a 4-H proton (9.13 ppm) and a quartet of the protons of the p-chlorophenyl ring in its PMR spectrum.

The structures of the substances obtained were confirmed by the IR and UV spectra and the results of elementary analysis.

LITERATURE CITED

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